

Science Highlights

from the National Synchrotron Light Source

BEAMLINES

X19A, X26A, X23A2

PUBLICATION

A. Leri, M. Hay, A. Lanzirotti, W. Rao, and S. Myneni, "Quantitative Determination of Absolute Organohalogen Concentrations in Environmental Samples by X-ray Absorption Spectroscopy," Analytical Chemistry, 78, 5711-5718 (2006).

FUNDING

Office of Basic Energy Sciences - U.S. Department of Energy; National Science Foundation (NSF) Chemical Sciences Program; NSF Graduate Research **Fellowships**

FOR MORE INFORMATION

Alessandra Leri, Department of Geosciences, Princeton University aleria@princeton.edu

Halogens undergo complex biogeochemical transformations in terrestrial systems, cycling between organic and inorganic forms. Naturally produced organohalogens include a diverse array of relatively low molecular weight molecules, as well as macromolecules of indeterminate structure. Despite the ubiquity of natural organohalogens, the processes associated with their formation and degradation in the environment remain poorly understood, largely due to the inadequacy of

available techniques to account for the myriad halogen species in heterogeneous soil, sediment, plant, and aqueous samples. Established quantitative methods require chemical isolation of organohalogen fractions from natural samples and are prone to partial recoveries and/or chemical alterations of halogens. We have applied synchrotron-based x-ray fluorescence (XRF) and x-ray absorption near-edge structure (XANES) spectrosQuantitative Determination of Absolute Organohalogen Concentrations in Environmental Samples by X-ray Absorption Spectroscopy

A.C. Leri, M.B. Hay, A. Lanzirotti, W. Rao, and S.C.B. Myneni 5,6

¹Department of Chemistry, Princeton University; ² Department of Civil and Environmental Engineering, Princeton University; ³Consortium for Advanced Radiation Sources, The University of Chicago; 4 Savannah River Ecology Laboratory, University of Georgia; 5 Department of Geosciences, Princeton University; ⁶Earth Sciences Division, Lawrence Berkeley National

A method for quantification of organic and inorganic halogen concentrations in environmental samples using x-ray absorption spectroscopy has been developed. Organochlorine ($Cl_{\rm org}$) and inorganic chloride ($Cl_{\rm inorg}$) concentrations are determined from Cl 1s XANES spectra. The absolute fluorescence intensity of these spectra is linearly dependent on CI concentration. Calibration curves are obtained from NaCl standards in a matrix of uniform bulk density, and Cl concentration in natural samples is estimated from these curves with high precision. The fractions of Cl_{inorg} and aliphatic/aromatic Cl_{ora} contributing to the total Cl 1s XANES signal are computed through least-squares spectral fitting with representative model compounds. Concentrations of organic and inorganic Br (Br organic and Br_{inorg}) in sediment samples are also measured using a combination of XRF and Br 1s XANES spectroscopy.

copy to the in situ determination of $\mathrm{Cl}_{\mathrm{org}}$ and $\mathrm{Br}_{\mathrm{org}}$ concentrations in natural samples.

Ouantification of CI relies on the absolute fluorescence intensity of Cl 1s XANES spectra (acquired at beamline X19A) determined at 2850.8 eV, an energy ~20 eV above the CI K-absorption edge (Figure 1A). By this energy, the near-edge oscillations are attenuated and absorption intensity is independent of CI speciation. Polyacrylic acid-based NaCl stan-



Graduate student author Alessandra Leri (left) and Michael Hay (right) collect natural organic matter samples in the field.

dards yield calibration curves with a strong linear relationship between CI concentration and absolute CI fluorescence intensity (Figure 1B). X-ray beam absorption and scattering by polyacrylic acid-based standards and natural organic matter (NOM) are comparable, allowing for direct measurement of CI concentrations in homogenized NOM sample pellets, which minimizes matrix effects.

Unnormalized Cl 1s XANES spectra of oak leaf NOM at progres-

sive stages of decay are shown in Figure 2A (a = least degraded; c = mostdegraded). Comparison of the absolute CI fluorescence intensity values of these spectra with standard curves (as in Figure 1B) yields the following total CI concentrations: a, 364 ppm; b, 141 ppm; c, 70 ppm. The decreasing trend in CI concentration is consistent with leaching of soluble forms of CI from the mulch material as part



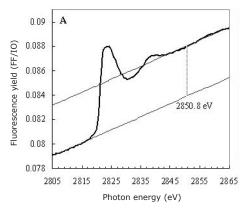
of the degradation process.

In addition to providing Cl concentration measurements, Cl 1s XANES spectra illuminate the chemical forms of CI present in a sample. Cl_{inorg} exhibits a broad absorption maximum (**Figure 2B,** x) higher in energy than the more intense maxima characteristic of Cl_{ara} compounds. These sharp peaks denote 1s \rightarrow π^* or σ^* transitions and differ in energy depending on C-Cl bond length, as becomes evident through comparison of aliphatic and aromatic Clora spectra (Figure **2B**, y-z). The substantial variations in spectral features depending on the coordination environment of CI allow percentage estimates of Clinora and aliphatic and aromatic Cl_{org} in natural samples to be ascertained via least-squares fitting of normalized sample spectra with spectra of representative model compounds (**Figure 2B**). Combination of these proportions with total Cl concentrations yields the concentrations of Cl_{inorg} , aromatic Cl_{org} , and aliphatic Cl_{org} in NOM samples (**Figure 2**, **caption**).

Total Br concentrations in environmental samples have been determined via XRF analysis for decades. The novelty of our application lies in the combination of quantitative information from XRF (acquired at beamline X26A) with Br speciation information from Br

1s XANES spectra (acquired at beamline X23A2) to determine absolute concentrations of $\mathrm{Br}_{\mathrm{org}}$ and $\mathrm{Br}_{\mathrm{inorg}}$ in chemically heterogeneous sediments.

This method provides a rigorous new approach to the determination of organohalogen concentrations in complex natural samples in situ. Robust, quantitative information about organohalogen fluxes in soil and sediment systems will help complete the description of natural halogen cycles, potentially illuminating the sources and sinks of naturally and industrially produced organohalogens in the environment.



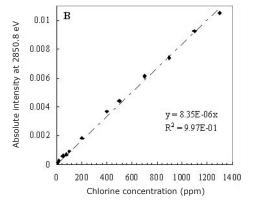
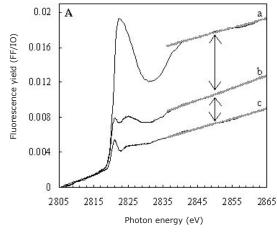


Figure 1. Quantification of CI concentration from CI 1s XANES spectra. A) Unnormalized CI 1s XANES spectrum of NaCl standard in PAA matrix (200 ppm CI). The difference at 2850.8 eV between splines through the pre-edge and post-edge regions is used as a measure of absolute CI fluorescence intensity. B) Relationship between CI concentration in PAA-based NaCl standards and absolute CI fluorescence intensity. Dashed line, equation, and R² value represent linear fits to the data.



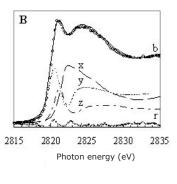


Figure 2. Determination of CI speciation and concentration in NOM samples from CI 1s XANES spectra. A) Unnormalized CI 1s XANES spectra of pulverized oak leaves at different stages of decay (a = least degraded; b = more degraded; c = most degraded). Arrows illustrate the differences in absolute CI fluorescence intensity among the spectra. B) Least-squares fit of background-subtracted, normalized spectrum b with representative inorganic (x, glycine-HCl), aliphatic (y, chlorodecane), and aromatic (z, chlorophenol red) CI model compounds. Spectrum b data are represented by circles and fit by the solid line; r is the residual after the fit. Estimated CI speciation by least-squares fitting: a, 78% inorganic (284 ppm), 16% aromatic (58 ppm), 6% aliphatic (22 ppm); b, 40% inorganic (56 ppm), 20% aromatic (29 ppm), 40% aliphatic (56 ppm); c, 16% inorganic (12 ppm), 42% aromatic (29 ppm), 42% aliphatic (29 ppm).